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WATSON'S PECANWOOD PLANTATION

WHITEFIELD W. WATSON
LARGE THIN SHELL PECANS
ORANGEBURG, S. C.
P. O. BOX 666.

CAROLINA "QUALITY" SOFT SHELL PECANS

While yield of our groves this year is disappointing, the quality (flavor, coloring and size) is unusually superior. Harvesting and curing process is progressing satisfactorily under favorable weather conditions and shipments will commence about 15th November. We hesitate to send announcements for fear of being oversold, but it is our desire first to take care of our regular customers as far as possible. We suggest that reservations be made early.

PRICE LIST

✓
"Stuart": This has proven to be the favorite of all pecans for general use, price considered, it being a large, handsome soft shell nut, full of rich meat and easily cracked and extracted: 2 pounds \$1.50; 5 pounds \$3.25; 10 pounds \$6.25; 50 to 100 pounds @ 60c per pound. Cartons sealed sanitary and insured.

✓
"Schley": This is the very thin "Paper Shell" nut regarded by some epicures as the choicest of all pecans for table use, being very rich meat and delightfully flavored. Quantity being limited the price is \$1.00 a pound in 1¼, 2½, 5 and 8 pound cartons.

Transportation: Prices include prepayment of postage, express and insurance of safe delivery.

C. O. D.: Cartons may be shipped 'Collect' if preferable.

Gift Packages: may be sent direct from us with cards.

Reference, by permission, Orangeburg National Bank.

W. W. WATSON
WATSON'S PECANWOOD PLANTATION
ORANGEBURG, SOUTH CAROLINA

THE UNIVERSITY OF CHICAGO
DEPARTMENT OF CHEMISTRY
RESEARCH REPORT
NO. 1000
1955

The following is a summary of the results of the experiments conducted during the past year. The work was carried out in the Department of Chemistry, University of Chicago, under the direction of Professor [Name]. The experiments were designed to investigate the properties of [Subject] and to determine the effect of [Factor] on the reaction rate. The results of the experiments are shown in the following table:

Run	Time (min)	Concentration (M)	Rate (M/min)
1	10	0.1	0.01
2	20	0.2	0.02
3	30	0.3	0.03
4	40	0.4	0.04
5	50	0.5	0.05

The results of the experiments show that the rate of reaction increases with increasing concentration of [Subject]. The rate of reaction is also affected by the temperature of the reaction mixture. The rate of reaction increases with increasing temperature. The results of the experiments are consistent with the theory of [Theory].

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